

Structural and Optical properties of Polymers(PVA / PVP) doped with V₂O₅ composites films

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Abstract

Study the effect of doping V₂O₅ on polymers poly vinyl alcohol (PVA), poly vinyl pyrrolidone (PVP) on the optical and structural properties for film prepared by using Casting method at thickness(300±20)nm ,All the materials dissolved in distilled water by magnetic mixer for one hour .The optical parameters measured by using UV-VIS spectrometer ,and the structural parameters measured by X-ray diffraction .when measured the energy gap found that the value was decreases from 4.6 eV to 2.98 eV with doping .The refractive index ,extinction coefficient ,absorption coefficient ,real and imaginary dielectric constants of (PVA/PVP) are increasing with doping by V₂O₅ and with increase the ratio of doping .

Key words: PVA , PVP, doping ,structural and optical properties,V₂O₅

Introduction

Composites of Polymer have been become on .account of simple process ability ,relatively in expensive ,very pest quality surface and simple fabrication of thin and thick samples [1,2].There are several factors which affect polymer properties such as optical ,structural ,electrical and doping ,It is possible to control the type of polymer through control of concentration ,interplay with the polymer chains doping [3].Interest in the study of electronic transitions of polymers gives a clear idea about the optical properties and that knowledge useful in many applications including protective coatings ,optical filter, Semiconductor and a cover Solar collection [2,4].

Poly (vinyl pyrrolidone)(PVP) and poly (vinyl alcohol)(PVA) are possess distinctive qualities from arrange of polymers are the possessing high polar ,simple processability and friends of environment ,And also characterized Superb thermo stability ,good electrical properties ,a loft mechanical strength and water Solubility [6,7].

From previes studies ,which focused on composites PVA/PVP doping are [8]."that they concluded for their studies polymer PVA and PVP thin films deposited onto SAW substrates by using spin coating .all samles show high sensitivity to relative humidity with range (5-95)%RH.and the other result that the better parameters of the PVP-senser" ."The Shen Hong ,C.Bolin et al [9] studied colloidal Au with PVP composite films prepared by spin-coating method ,then found that large optical nonlinearities of Au/PVA films".Poly (vinyl pyrrolidone)PVP andPoly(vinyl alcohol)PVA,are in the list of synthetic polymers which are used in medicine.[4].Vanadium Pentoxide is used in many applications such as gas sensing agent , a window for solar cells and the important application of V_2O_5 is the positive material of lithium ion batteries..[5]

The aim of the work preparation of (PVA-PVP- V_2O_5) films and study their structural and optical properties.

Method and Materials

In this research ,putting 4gm of PVA with ($w_t=30000$ gm/mol) and 4gm of PVP with ($w_t=70000$ gm/mol) dissolved in 100mil distilled water by magnetic mixer for one hour , NH_4NO_3 dissolved in distilled water at concentration 2% and 6% by magnetic mixer for one hour .Two dropes of each solution is put in glass petri deshes at dimeter (4cm) and left to dry in the room temperature for 72 hour ,the samples thickness were (300) nm ,which was measured by weighting method depending on the equation[6].

$$th = \Delta m / \rho a \quad \dots \dots \dots (1)$$

when th:is thickness , Δm : the different between weight substrate before and after the film deposition , ρ : the density, a:is the area of the film . The transmission measured by (UV-visible spectroscopic effect) As well as sampling X-ray diffraction by shemadzu XRD – diffractometer . The structural parameters was calculated such as :

the Grain Size [10]. By Scherrers equation can be calculated

$$G.S = M \lambda / \beta \cos \theta \quad \dots \dots \dots (2)$$

Where M is constant equal to (0.94), λ is the wavelength of X-ray (1.5406 \AA) , β is the full width at half maximum (FWHM) and θ is the Bragg's angle .

Using The dislocation density (η) also calculated by "

Williamson and Smallmans equation [11]".

$$\eta = 1 / (G.S)^2 \quad \dots \dots \dots (3)$$

Also the number of crystals (N_c) calculated by using the equation .[12].

$$N_c = th / (G.S)^3 \quad \dots \dots \dots (4)$$

At the end of structural parameters the strain (S) calculated by the formula . [13]

$$S = \beta \cos \theta / 4 \quad \dots \dots \dots (5)$$

All these parameters put in the table (1) The value of the number of crystals and dislocation density and the strain was increase at doping ratio 6% .

The energy band gap (Eg) could be calculated by equation . [14] and show in shape (4) .

$$\alpha h\nu = L(E-E_g)^J \dots\dots\dots (6)$$

where α is the absorption coefficient , ν is the incident photon frequency, L is constant and h is planck constant ,if J equal (1/2) the allowed direct transition ,but if equal to (2) the allowed in direct transition , the value of Eg recorded in table (2)

The energy E can be calculated by the relation . [15]

$$E = h\nu = 1240/\lambda \dots\dots\dots (7)$$

The reflectance R also calculated by the formula . [16]

$$R = 1 - (A + T) \dots\dots\dots (8)$$

The extinction coefficient (K_o) can be calculated by the equation .[17] and show in shape (5).

$$K_o = \alpha\lambda / 4\pi \dots\dots\dots (9)$$

After calculated R and K_o can be easily to calculated refractive index by the relation [18].and draw in fig (6) .

$$n = \frac{1+R}{1-R} + \left(\frac{4R}{(1-R)^2} - K_o^2 \right)^{1/2} \dots\dots\dots (10)$$

The dielectric constant into two types ; real (ϵ_r) and imaginary (ϵ_i) can also evaluated by the relation [19].and show in figure (7,8) .

$$\epsilon_r = n^2 - k_o^2 \dots\dots\dots (11)$$

$$\epsilon_i = 2nk_o \dots\dots\dots (12)$$

The optical conductivity as a function of photon energy can be also calculated by the relation [20].and show in shape (9).

$$\delta = \frac{\alpha nc}{4\pi} \dots\dots\dots (13)$$

Where c: the velocity of light .

By using the relation no. (13) can be found the absorption coefficient [21].and show in figure (10).

$$\alpha = 2.303(A/th) \dots\dots\dots (14)$$

Discussion of Results

Figure (1) show the XRD for PVA , PVP , PVA:PVP:V₂O₅(2%) and PVA:PVP:V₂O₅ (6%) . we found that the samples of V₂O₅ is orthorhombic structure , then the peaks of [600] plane at 37.8° , [002] plane at 39.08° and [020] at 37° .

The optical parameters was studied like the spectra of absorption A% and transmission as shown in shape (2-3) we note the absorption spectra of the samples as a function of wave length decrease with increasing of doping ratio. This behavior due to absorbs the incident light depit the V₂O₅ particles ,opposite the transmission which is increases with increase the doping ratio from 2% to 6% then is very high equal to \cong (95%) from these samples that is good to use in solar cell and other applications.

The optical band gap (Eg) which is clear in shape (4) decreases with doping in V₂O₅ and with increasing the ratio of doping ,this decrease of band gap due to the decrease in crystal size of original solution ,(Hussein 2015).

The observe increment of extinction coefficient in figure (5) at high wave length with doping and increasing of doping ratio of V₂O₅ due to more scattering of photons (Tintu 2010) [3] .

In the shape (7,8) we observed that ϵ_r behaves just like refraction index and extinction coefficient because they joined with it in relations .the real part of dielectric constant as a function of photon energy is increase at doping with V₂O₅ because increase the absorption of incident light .(Ahmed 2007).[22].

.we observed the optical conductivity was increase with increases of photon energy and this behavior is same as the absorption coefficient because it depending on it by the relation number (13) .

In the figure (10) we show the increase of absorption coefficient as a function of photon energy with increasing the ratio of doping because increase number of carries charges .and because the absorption coefficient less than 10^4 so the band gap is indirect . .

References

- 1- Rawat , A., Mahavar, H.K. ,S.chauhan ,A.Tanwar and P.J.Singh,(2012) "optical band gap of polyvinyl prrolidone /poly acrilamide blend thin films "J.pure &Applied physics .:50, (100-104) .
- 2- Hahim, H. ,Khalaf, A.,Ghazi, J.H Hashim, A.,and M.Khudhur,(2014), "optical properties of poly vinyl prrolidone (pvp) Films doped with Nickel Nitrate (NiNO_3) ,J.Science and Research (IJSR) ,.3,No.10, (1869-1870) .
- 3-.Hakim.H.,Al-Ramadhan Z andHashim A, (2015),"The Effect of Aluminum oxide Nanoparticles on the optical properties of (PVP-PEG) Blend ", J.Celland Molecular Biology ,:5,No.4, (15-18) .
- 4-Ibrahim R.A gool and A.Hashim,(2014), "Enhancement of structural and optical properties of (PVA-PVP- TiO_2) nano composites)..8,No.17, (564-570).
- 5-Hassan.K, Asieh.M ,(2015),"Poly vinyl Alcohol-Based Sol-Gel synthesis of V_2O_5 Nanoflakes as Positive Electrodes of Li-ion Batteries .Int.J.Electrochem.sci.,10.7392-7408.Iran.
- 6- Premalatha, M, A Vijaya.N.,Selvalakshmi. S and.Monisha. S. (2015),"study of proton conducting Blend polymer Electrolyte Based on PVA-PVP" j.Medical science ,.4 ,No.6, (189-192).
- 7-Dinkhan, S.,Ahmed. B , Raghuvanshi, S.K, Wahab, M.A. (2014), "structural ,morphological and optical properties of silver doped polyvinyl prrolidone composites " J.pure &Applied physics ,.52,.(192-197) .
- 8-.Buvailo . A.,Xing . Y.,Hines .J E.Borguet,(2011), "Thin polymer film based rapid surface acoustic wave humidity sensors "J.Sensors &Actuators , .156 . (444-449).
- 9- Kadhim , R.G,(2014),"Effect of Nano silver particles on the optical and structural properties of (PVA-PVP)Films" J .Advances in physics Theories and Application,,34,.(30-39).
- 10- Hong, S. Bo-lin C, wei.L.G. Tian.W.W. G.D.Yi,C.Z.Hao,and Zhen Y.G. ,(2005) ."Nonlinear optical properties of Au/pvp composite thin films .J.Chinese physics and IOP Publishing Ltd ,.14,No.9, (1915-1918).
- 11-Jain. A .P.Sagar ,R.M. (2007),"change of structural ,optical and electrical properties of sol – gel derived Zno films with their thickness ",J.Materials science –Poland ,.25,No.1, (233-242) .
- 12-Ezema. F.L, (2004), " Fabrication ,optical properties and Applications of undopel chemical Bath deposited Zno thin films ",J. science ,.15,No.4 , (343-350) .
- 13- Sub E, shim , .Seong. H,S.Sik , Yan. I,and S.Yeal,(2003), "Annealing effect on the structural and optical properties of Zno thin films on Inp" ,J.Materials science and Engineering , B102 ,.(366-369).
- 14-Cullity B.D,(1956),Elements of X-ray Diffraction ,Addison –Wesley publications company Inc, Reading ,Massachusetts .
- 15- Tanc, J. (1974)."Amorphous and liquid semiconductors" ,plenum prees ,New York .
- 16- Noweell MM,Pandel NR ,and Liu x,(2015)," characterization of sputtered CdTe thin films Electron bactter diffraction and correlation with device performance " ,J.Micosc Microanal ,.21, 4, (27-35).
- 17- Divya.R.,M.Meena ,C.K. Mahadevan and Padma, C.M.,(2005),"Investigation on Cuo Dispersed PVA polymer films ".4.(2248-9622).

- 18-Caglar M. Caglar .Y., S.I lican ,(2006)," The determination of the thickness and optical constants of the ZnO crystalline thin film by using envelop method " ,J. optoelectronics and Advanced materials , .8.4 . (1410-1413)
- 19- Aksoy. S. . Caglar ,Y,S. I lican ,(2010),"Effect of Sn dopants on the optical and electrical properties of ZnO films " ,J.optica Applicata ,.xl,No.1, (7-13) .
- 20- Al-Douri ,A.A.,Al-Shakily ,F.Y.,Alia M.F and Al najjar A.A.(2010),"optical properties Al and Sb doped CdTe thin films " J.Advances in condensed Matter physics ,.10, (1-5) .
- 21- Buet. F, J.Olivier Fourcade ,Y.Bensimon ,P. Belougne ,(1991), state communications ,77(1).(29-32) .
- 22- Mustafa. Mustafa F.A.,(2013) ,"optical properties of NaI doped poly vinyl alcohol films " ,J.physical science Research International ..1(1), (1-9).
- 23-Ahmed, A.H. Awatif .A.M and N.ZeidAbdul-Majied , 2007."J.ofEng.&Technology" ,25(4):558-568.

Table (1): The structural parameters from grain size ,dislocation density ,number of crystals and strain .

| Sample | G.S(nm) | $(nm)^{-2}$ | $N_0 (nm)^{-2}$ | S |
|--|---------|-------------|-----------------|--------|
| PVA | 0.8362 | 1.4302 | 513.17 | 0.4143 |
| PVP | 0.4690 | 4.5475 | 2909.79 | 0.7387 |
| PVA:PVP:V ₂ O ₅ (2%) | 0.7727 | 1.675 | 650.33 | 0.4484 |
| PVA:PVP:V ₂ O ₅ (6%) | 0.3273 | 9.337 | 8571.42 | 1.058 |

Table (2): The value of energy gap from the samples

| Sam. | Eg(eV) |
|--|--------|
| PVA | 4.6 |
| PVP | 4.42 |
| PVA:PVP:V ₂ O ₅ (2%) | 3.49 |
| PVA:PVP:V ₂ O ₅ (6%) | 2.98 |

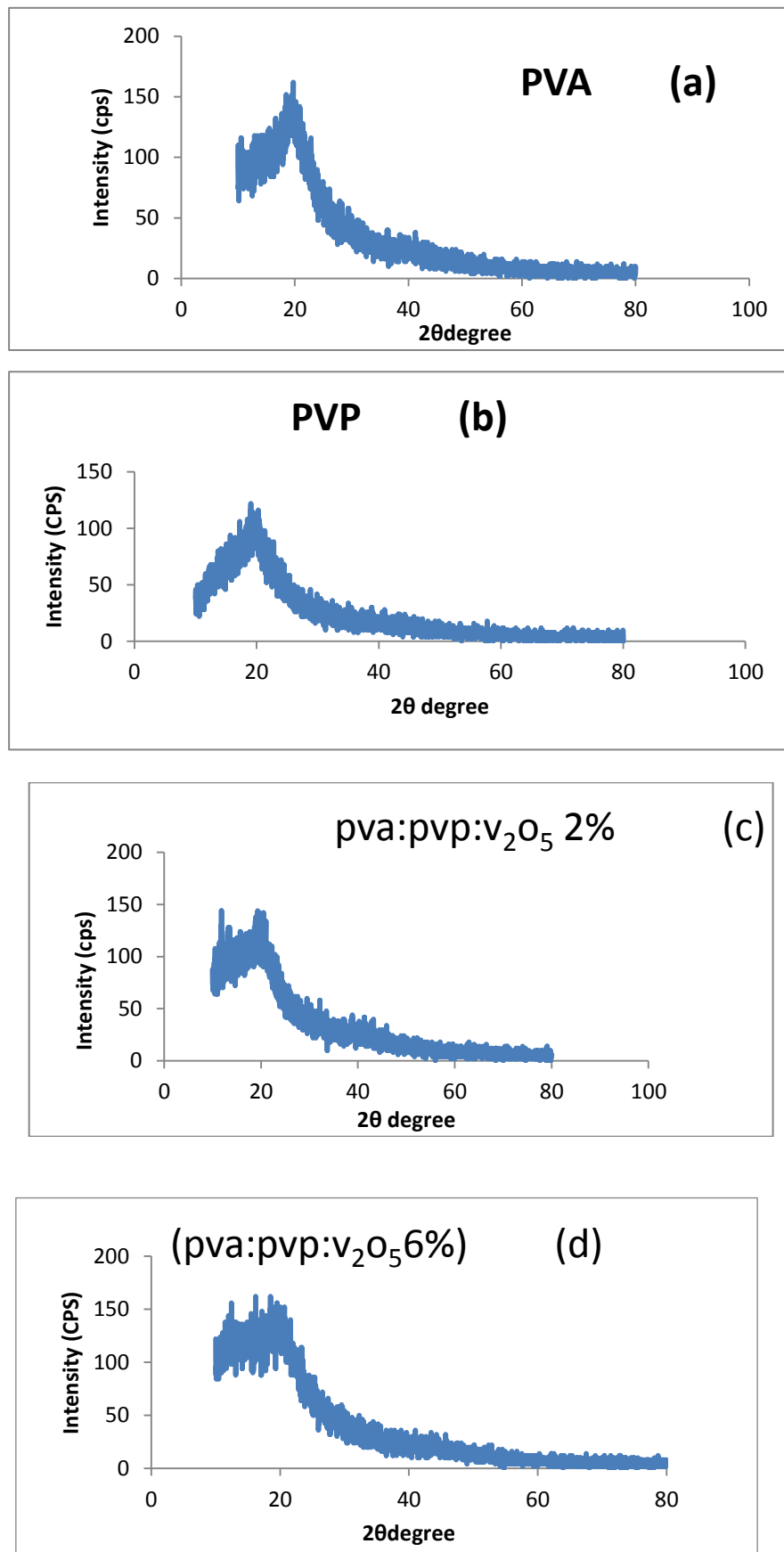


Figure (1) :(a,b.c.d) X-ray diffraction for the thin film.

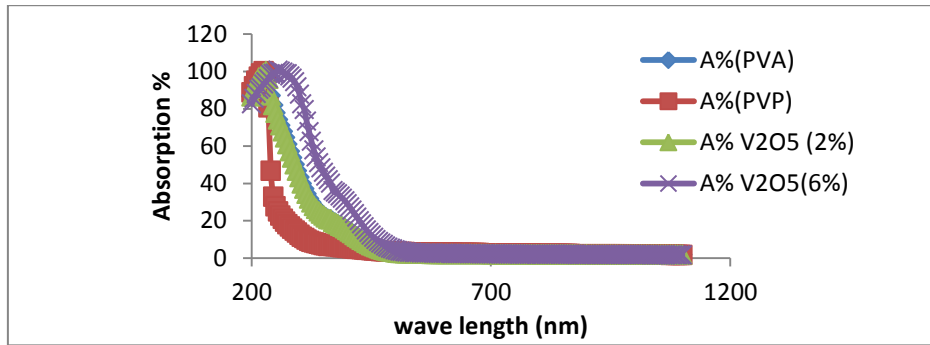


Figure (2):The Absorption ratio as a function of wavelength

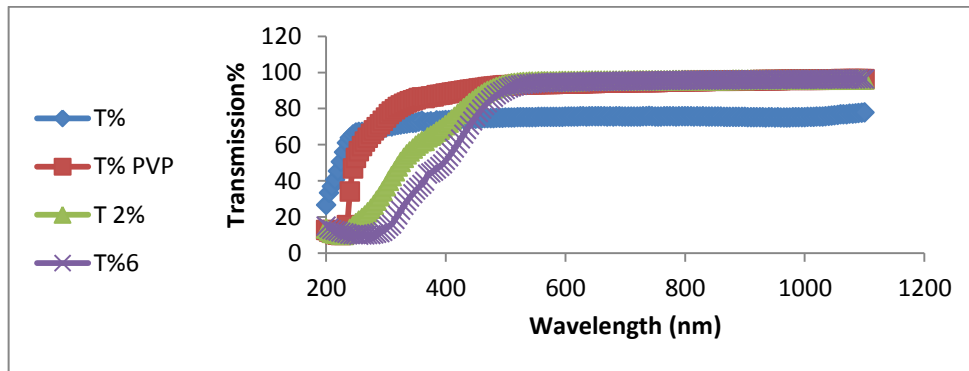


Figure (3): The Transmission ratio as a function of wavelength

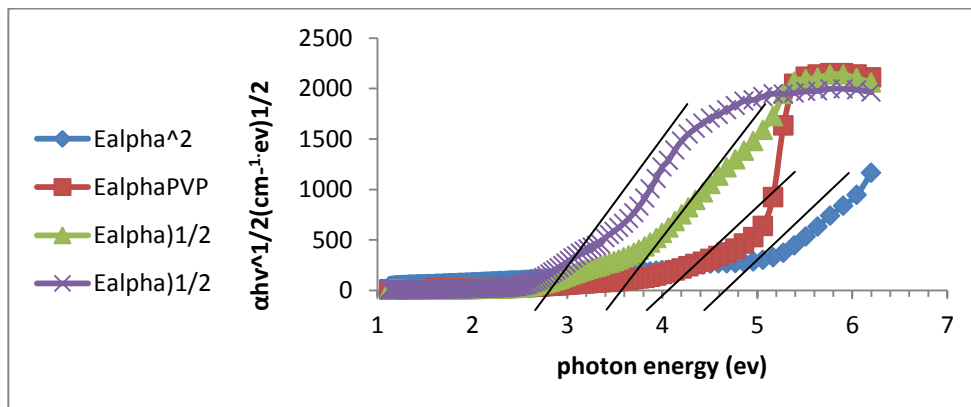


Figure (4):The relation between $(\alpha h\nu)^{1/2}$ and photon energy .

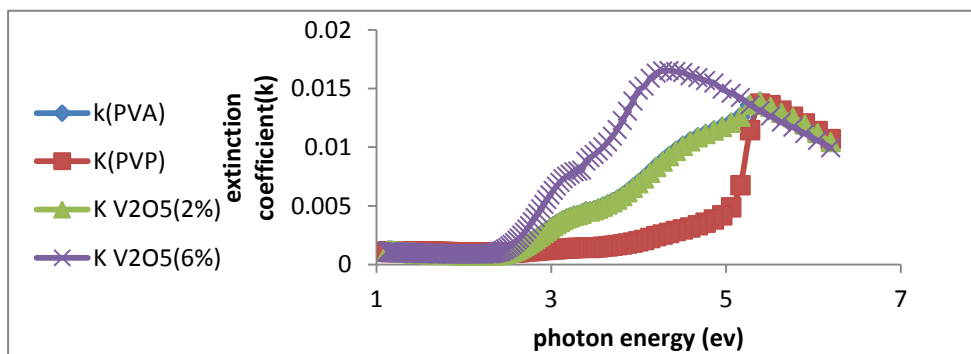


Figure (5): The Extinction coefficient as a function of photon energy.

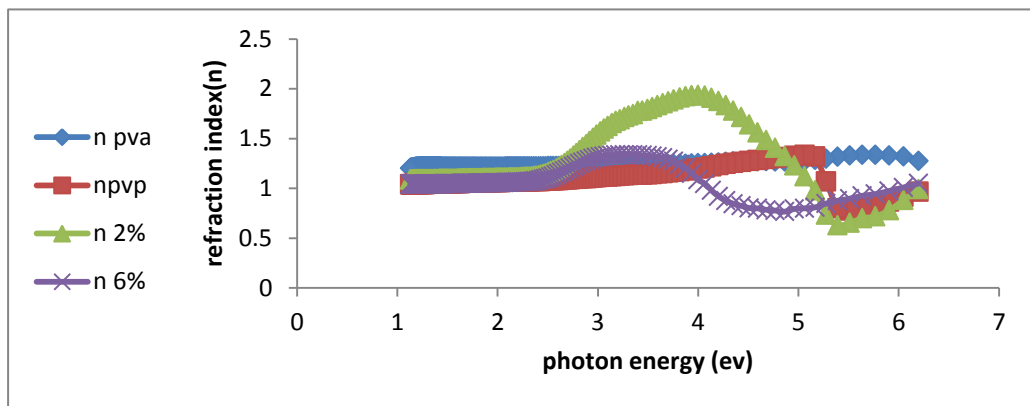


Figure (6) :The refractive index as a function of photon energy

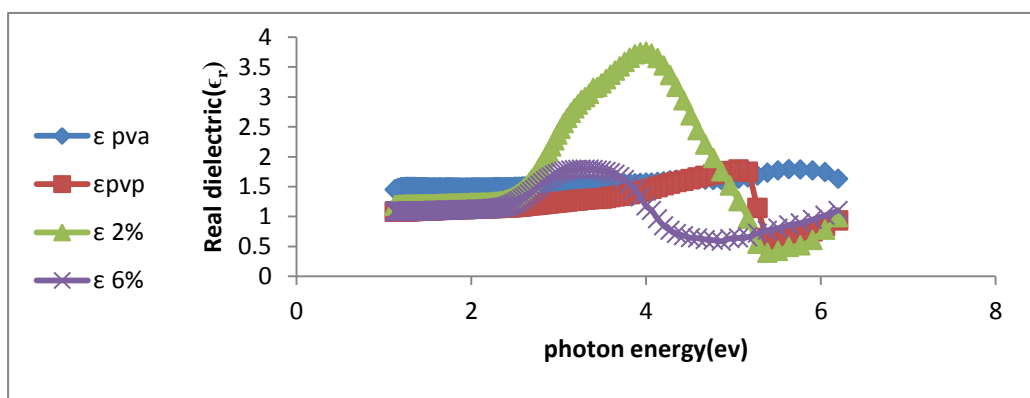


Figure (7): The real dielectric as a function of photon energy

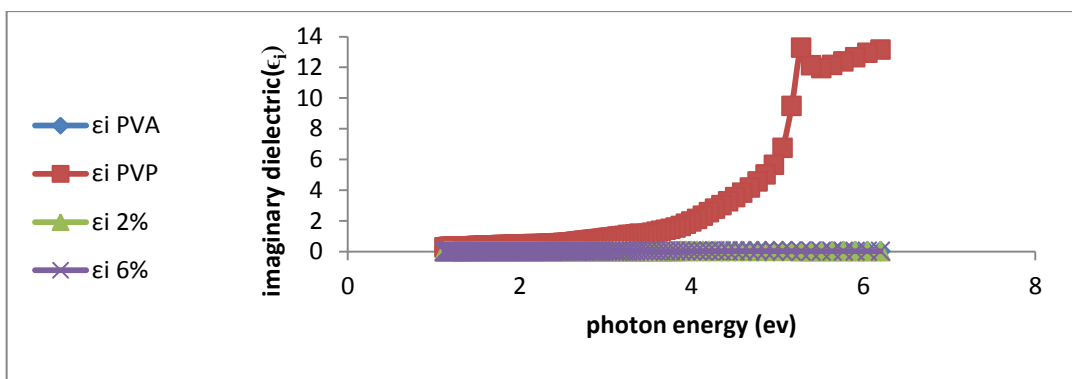


Figure (8):The imaginary dielectric as a function of photon energy .

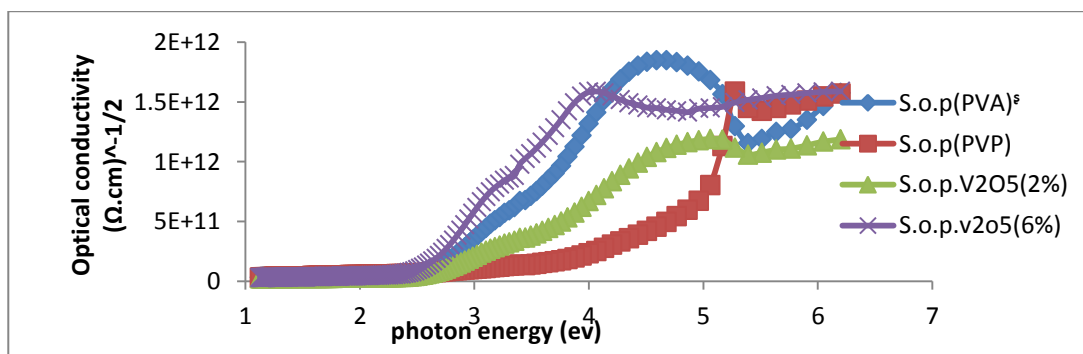


Figure (9):The optical conductivity as a function of photon energy .

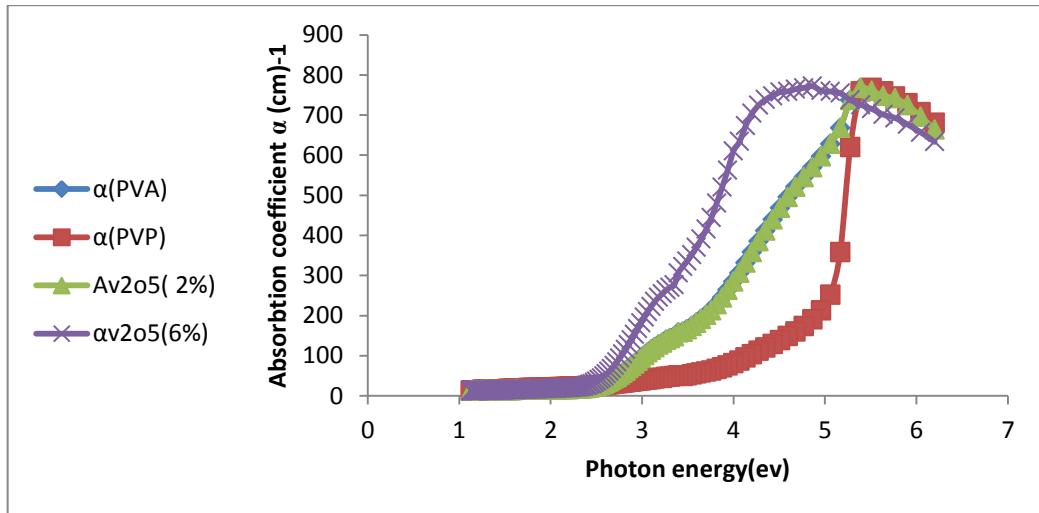


Figure (10):The Absorption coefficient as a function of photon energy.