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# **Charge Transport in Magnetized Plasma**

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### Abstract

The plasma source can restrict the motion of charges that are localizing in the non equilibrium distribution of charge energy and reducing the electrons transport across magnetic field. The electrons & ions motion are controlled by ambipolar electric field and charge-atom collision. The source density for a given electron temperature and a given ion are considered to evaluate the diffusion coefficient. The ambipolar diffusion coefficient and the cross field diffusion coefficient for charge transfer are calculated through magnetized plasma in a uniform magnetic field, and an approximation ambipolar diffusion coefficient is evaluated. The result, showes how the diffusion process is gradually imbedded as the properties of the plasma.

Key words :- Change transport, monetized plasma, diffusion coefficient

#### Introduction

The transfer of wave function energy towards the long region and the formation of longe scale structures is a result of the well-known inverse cascade in two dimensional and quasi two – dimensional fluids [1]. A complete understanding of charge transport is important for characterizing materials used in the components which are directly exposed to the charge (proton) isotope plasma [2]. The source of electron transport in magnetized plasmas which can be a major obstacle in the way toward particles nuclear fusion power. The observed electron energy transport is much larger than are would expect from diffusion process due to coulomb collisions [3]. The diffusion has important technological implications in micro electronics [4]. Coalitional cross field transport due to electric or magnetic field asymmetries is important in many neutral and non neutral plasma confinement devices [5]. It is the purpose of this paper to give transport description of diffusion across magnetic field in plasma source.

# Theory

Transport in the discharge is controlled by equilibrium magnetic field, ambipolar electric field, and ion – atom collisions [4 - 5].

For a plasma with a single species of singly charged ions, the am bipolar diffusion coefficient in a weakly ionized system of sufficient size is given by [6].

$$D = \frac{c_{s(1+T_i/T_e)}}{n_n (\sigma_e \sqrt{m_e/m_i} + \sigma_i \sqrt{T_i/T_e}} \quad \dots \quad 1$$

Ibn Al-Haitham Journal for Pure and Applied Science						بقية	ة و التطبي	م الصرف	الهيثم للعلو	مجلة إبن		
No.	$\boxed{1}$	Vol.	$\boxed{25}$	Year	2012	7.9	2012	السنة (	25	المجلد (	$\left(\begin{array}{c}1\end{array}\right)$	العدد

Where  $n_n$  is the neutral gas density,  $\sigma_e$  and  $\sigma_i$  are the total scattering cross section with neutrals for electrons and ions respectively,  $m_e$  is the electron mass,  $m_i$  is the ion mass,  $T_e$  is the electron temperature,  $T_i$  is the ion temperature and  $c_s$  is the ion sound speed [7]. is given by [6-7].

$$c_s = \sqrt{k T_e/m_i} \qquad \dots \qquad (2)$$

The cross section for electrons or ions are given [8]

Where  $\lambda$  is the collisonal mean free path .

The ion – atom collision frequency is given by [6]

$$v = n_o v \left( \sigma_e + \sigma_i \right) \qquad \dots \qquad (4)$$

..... (3)

When  $n_0$  is the gasses density and v is the ion velocity is given by

 $\sigma = 1/n\lambda$ 

$$\boldsymbol{v}_i = \sqrt{\frac{2T_i}{m_i}}$$

However when the electron – ion mass ratio  $\ll T_e/T_i$  and small value that can be ignored in Equ(1) and results .

$$\mathbf{D} = \frac{c_s}{n_n \sigma_i} \sqrt{\frac{T_g}{T_i}} \qquad \dots \dots (5)$$

The magnetic field can inhibit electron motion perpendicular to the magnetic field lines described by cross field diffusion coefficient  $D_{\rho}$  and given [7].

$$D_{\rho} = \frac{n_n \sigma_e v_{th}}{2} \rho_e^2 \qquad \dots \dots (6)$$

(	Ibn Al-Haitham Journal for Pure and Applied Science					يقية	ة و التطب	م الصرف	الهيثم للعلو	مجلة إبن	•	)		
	No.	$\boxed{1}$	Vol.	25	Year	2012	π.	2012	السنة	25	المجلد (	$\left(\begin{array}{c}1\end{array}\right)$	العدد	

Here  $\Psi_{th}$  is the electron thermal speed and  $\rho_e$  is the electron gyro radius is given by [6]

$$\rho_{\rm e} = \frac{v}{w} = \frac{n_{\varrho} \quad v \quad (\sigma_{\varrho} + \sigma_{i})}{w} \qquad \dots \dots (7)$$

Whene the plasma frequency is given by [9].

Herne B is the static magnetic field strength

#### Results

In order to determine the diffusion coefficient of charge transport in magnetized plasma theoretically using the equation (1), one must initially evaluate the values of the ions sound speed  $c_s$  form equation (2) for a variety ions Hydrogen, Argon, and Nitrogen where the energies of electrons kT<sub>e</sub> taken between (1 to 2.4) ev [10]. The values of mass of ions are m<sub>H</sub> = 1.67826\*10<sup>-27</sup>, m<sub>ar</sub>= 2.67\*10<sup>-26</sup> kg, and m<sub>n</sub> = 2.5\*10<sup>-26</sup> kg were extracted from the literature [11-12].

Amore general expression equation (2) was applied to evaluate the sound speed of Argon , Hydrogen , and Nitrogen cons ions with masses of these ions , the results have been summarized in table (1).

We use the results of sound speed ions  $c_s$  in table (1) to calculate the diffusion coefficient charge stimulated by plasma by using equation (1) with values of  $T_i = 0.1$  ev [7],  $n_a = 7.2*10^{19}$  m<sup>-3</sup> [13], and  $\sigma_i$ ,  $m_i$ , and  $m_e$  from table (2), the results are tabulated in table (3).

Another important parameter for diffusion is the overall ion – atom collision frequency  $v_i$  – atom that can be calculated from equation (4), where the gas density  $n = 7.2*10^{-19} \text{ m}^2$  and the values of v,  $\sigma_e$ , and  $\sigma_i$  are taken from table (2), the values of  $v_i$  – atom are summarized in table (4).

So the other variable in diffusion of charge transport is the collisonal mean free path can be evaluated by using equation (3) and  $\sigma_i$  from table (2), these calculated values are shown in table (5).

The diffusion coefficient of am bipolar that caused by the direct ion motion modify the am bipolar flow can be calculated by equation (5) with used value of  $c_s$  from table (1) and  $\sigma_e$ ,  $\sigma_i$ ,  $T_i$ , and  $T_e$  from table (2). results are summarized in table (6) also , the cross–field diffusion coefficient  $D_\rho$  that describe the electron motion perpendicular to the magnetic field can calculated by used equation (6) after estimated the value of electron gyro radius we estimate the transport properties of the magnetized electrons by used equation (7) and (8) the plasma frequency w can be estimated by using equation (8), the results are show in table (7).

The gyro radius of electron  $\rho$  can be evaluated when inserting values of w in equation (7) with values of V<sub>i</sub> from table (4) the results are summarized in table (8) . finally by using the results of  $\rho$  from table (7) with eq (6), we can calculate the cross field diffusion coefficient the results are listed in table (9).

Ibn A	Ibn Al-Haitham Journal for Pure and Applied Science						قية	فة و التطبي	لوم الصر	، الهيثم للع	مجلة إبن	
No.	$\left[ 1 \right]$	Vol.	25	Year	2012	7.9	2012	السنة (	25	المجلد	$\left(\begin{array}{c}1\end{array}\right)$	العدد

All results the ion sound speed  $C_s$ , diffusion coefficient D, ion-atom collisions frequency,  $\upsilon$ , collision and mean frequency path  $\lambda$ , ambipolar diffusion coefficient  $D_{\rho}$ , ions angular gyro frequency W, and ions gyro radius  $\rho_i$  are calculated using amatlab program.

# Discussion

For all the results reported here we consider the dimensions of the plasmas are small to justify a transport that relies on thermal equilibration of the electrons.

For the discharge considered here the ion temperature is expected to be reasonably close to the temperature of the neutrals roughly 0.1 ev.

The resulting values of the ion sound speed  $c_s$  was unusually high for nitrogen and argon comparing with low for hydrogen this indicate of  $c_s$  proportional with  $1/m_i$ .

Table (3) shows the overall diffusion coefficient are large for Nitrogen comparing with Argon and Hydrogen for all the same spectrum temperature value for (1to2.4) ev. This indicates that the diffusion coefficient is depending on the value of  $c_s$  sound speed of ion and the scattering cross section for ions that is very view in tables (1) and (2) respectively.

It turned out that the diffusion path way strongly depends on the cross section  $\sigma_i$ . Whereas the diffusion is favored in Nitrogen compare with other elements, that smean when  $\sigma_i$  small then mean path  $\lambda$  is Large and diffusion coefficient is Large and vice versa.

For electron temperature around (1 to 2.4) ev the ratio of  $T_i/T_e$  in equation (1) can be ignored [7] and the directed ion motion modify the am bipolar flow caused the am bipolar diffusion coefficient equation (5) the result of am bipolar diffusion coefficient indicates the diffusion in Nitrogen is more

active comparing with Argon and Hydrogen these depending on value of  $c_s$ ,  $\sigma_i$  and  $T_e$ .

Table (9) shows that the cross – field diffusion coefficient  $D_{\rho}$  that described the magnetic field can inhibit electron motion perpendicular to the magnetic field lines. The transport of electrons across magnetic field lines is affected by the magnetic field strength when cross – field diffusion of electron gyro – orbits becomes smaller than the am bipolar diffusion.

# Conclusions

In this work, the change transport in magnetized plasma source operation are studies in which the ions motion is controlled by the am bipolar electric field and ion – atom collisions. The sound speed of ion are calculated and found large values for nitrogen and mid large for argon and small for Hydrogen. the most large sound speed leads to height value of diffusion coefficient. in summary, the diffusion coefficients are calculated using equations (1), (5) and (6)

Showing large value for nitrogen compared with argon and hydrogen depending or co and speed of ion and scattering cross section for ions

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	No.	$\boxed{1}$	Vol.	25	Year	2012	2012

Table (1): The ion sound speed values for Hydrogen, Argon, and Nitrogen

T <sub>e</sub> (ev)		c <sub>s</sub> (m/s)	
-	H <sub>2</sub>	A <sub>r</sub>	N <sub>2</sub>
1	309.34	2447.96	8000
1.1	3244.44	2567.44	8390.47
1.2	338.80	2681.60	8363.56
1.3	352.70	2791.10	9121.40
1.4	366.02	2896.46	9465.72
1.5	378.86	2998.12	9797.95
1.6	391.29	3096.45	10119.28
1.7	403.33	3191.74	10430.72
1.8	415.020	3284.28	10733.12
1.9	426.40	3374.28	11027.23
2	437.47	3461.93	11313.70
2.1	448.28	3547.43	11593.10
2.2	458.83	3630.91	1186.91
2.3	469.14	3712.51	12123.60
2.4	479.23	3792.36	12393.54

Table (2): Velocity of Nitrogen, Argon, and Hydrogen ions and electron

ion	m <sub>i</sub> (Kg)	σ <sub>i</sub> (m²)[14]	T <sub>i</sub> (eV)[7]	V <sub>i</sub> *10 <sup>3</sup> m/sec
N <sub>2</sub>	2.5*10 <sup>-26</sup> kg	2.5*10-17 m <sup>2</sup>	0.1 ev	3.577708 m/s
A <sub>r</sub>	2.6*10 <sup>-26</sup> kg	3.4*10-17 m <sup>2</sup>	0.1 ev	1.094761103 m/s
H <sub>2</sub>	1.672*10 <sup>-27</sup> kg	3.9*10-17 m <sup>2</sup>	0.1 ev	4.374786393 m/s
Electron	9.1*10 <sup>31</sup> kg	5*10-19 m <sup>2</sup> [7]	1–2.4 ev	

Ib	n Al-Haitham	Journal for	Pure and	Applied Science	
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25

Year

Vol.

1

No.

المجلد

العدد

1

25

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Table (3): The ambipolar diffusion coefficient for Nitrogen, Argon, and Hydrogen ions

2012

T <sub>e</sub> (ev)	Diffu	sion coefficient D(m	$^{2}eV/Kg)^{1/2}$
	H <sub>2</sub>	A <sub>r</sub>	$N_2$
1	0.215491275	1.95580081	8.688021893
1.1	0.229542527	2.083315415	9.254354662
1.2	0.24327549	2.208376575	9.809789195
1.3	0.256843827	2.33110882	10.35484973
1.4	0.270142231	2.451753318	10.890642
1.5	0.283219488	2.570477453	11.41790119
1.6	0.296109499	2.687429174	11.9372269
1.7	0.309242861	2.802691618	12.44915585
1.8	0.321338248	2.916426174	12.95422078
1.9	0.333714949	3.028698493	13.45278679
2	0.345936101	3.13965136	13.94553699
2.1	0.358012275	3.249205352	14.4320167
2.2	0.369960597	3.357631219	14.91349599
2.3	0.381775697	3.4648666566	15.38971383
2.4	0.393478486	3.571073853	15.86132399

Gege of Table (4): Ion - atom collision frequency for Nitrogen, Argon, and Hydrogen ions

ion	$\sigma_{ion}(m^2)$	M <sub>ion</sub> (kg)	V <sub>Ar</sub> (m/sec)	υ <sub>ion</sub> (1/sec)
N2	2.5*10 <sup>-17</sup>	2.5*10 <sup>-26</sup>	3577.708764	656873.291
Ar	3.4*10 <sup>-17</sup>	2.67*10 <sup>-26</sup>	1094.761103	2719386.58
H2	3.9*10 <sup>-17</sup>	1.6726*10 <sup>-27</sup>	4374.786393	12441892.5

Ibn Al-Haitham Journal for Pure and Applied Science						يقية	ة و التطب	م الصرف	الهيثم للعلو	مجلة إبن		
No.	$\boxed{1}$	Vol.	25	Year	2012	Π <b>9</b> -	2012	السنة (	25	المجلد	$\left(\begin{array}{c}1\end{array}\right)$	العدد

Table (5):The collisional mean free path  $\lambda$  and cross section  $\sigma_{ion}$  for Nitrogen ,Argon , and Hydrogen ions

ion	$\sigma_{ion}(m^2)$	$\lambda_{ion}(m)$
N <sub>2</sub>	2.5*10 <sup>-17</sup>	5.55555*10-4
A <sub>r</sub>	3.4*10 <sup>-17</sup>	4.08496732*10 <sup>-4</sup>
H <sub>2</sub>	3.9*10 <sup>-17</sup>	3.561253561*10 <sup>-4</sup>

Table (6): The ambipolar diffusion coefficient approximation for Nitrogen,

TIL

Argon, and Hydrogen ions

T <sub>e</sub> (ev)	D:ambipolar diffusion coefficient (m <sup>2</sup> /sec)						
AT	H <sub>2</sub>	Ar	N <sub>2</sub>				
-1	0.348377135	0.399609066	0.54346833				
1.1	0.383208883	0.439563131	0.597805858				
1.2	0.417969552	0.479435662	0.652032501				
1.3	0.452876917	0.519476464	0.706487991				
1.4	0.487723575	0.559447631	0.755197372				
1.5	0.522551113	0.599396865	0.815179737				
1.6	0.557393162	0.639362745	0.869533333				
1.7	0.59222809	0.679320456	0.92387582				
1.8	0.62706468	0.719280075	0.978220902				
1.9	0.661908994	0.759246235	1.03257488				
2	0.696753844	0.799189444	1.086935997				
2.1	0.73158014	0.839165455	1.141265019				
2.2	0.766420863	0.879130185	1.195617052				
2.3	0.801252279	0.919083497	1.249953556				
2.4	0.836105966	0.959062751	1.304325342				

Ibn A	l-Haitha	m Journal	for Pure	and Applic	ed Science		يقية	ة و التطب	م الصرف	الهيثم للعلو	مجلة إبن	•	)
No.	$\boxed{1}$	Vol.	25	Year	2012	T.9	2012	السنة	25	المجلد	$\left(\begin{array}{c}1\end{array}\right)$	العدد	

Table (7): The ions angular	gyro frequency w(sec)	<sup>-1</sup> for Nitrogen, Argo	ı, and Hydrogen ions

			Magneti	c field streng	th B[15]		
ions	3*10 <sup>4</sup> Gaus	3.4*10 <sup>4</sup> Gaus	3.8*10 <sup>4</sup> Gaus	4.2*10 <sup>4</sup> Gaus	4.6*10 <sup>4</sup> Gaus	5*10 <sup>4</sup> Gaus	5.4*10 <sup>4</sup> Gaus
A <sub>r</sub>	1.797753*10 <sup>11</sup>	2.03745*10 <sup>11</sup>	2.277153*10 <sup>11</sup>	2.5168539*10 <sup>11</sup>	2.75655*10 <sup>11</sup>	2.99625*10 <sup>11</sup>	3.23595*10 <sup>11</sup>
H <sub>2</sub>	2.87425*10 <sup>9</sup>	3.257465*10 <sup>9</sup>	3.640718*10 <sup>9</sup>	4.02395209*10 <sup>9</sup>	4.4071856*10 <sup>9</sup>	4.790419*10 <sup>9</sup>	5.173652692*10 <sup>9</sup>
N <sub>2</sub>	1.92*10 <sup>12</sup>	2.176*10 <sup>12</sup>	2.432*10 <sup>12</sup>	2.688*10 <sup>12</sup>	2.944*10 <sup>12</sup>	3.2*10 <sup>12</sup>	3.456*10 <sup>12</sup>
					-		

Table ( 8 ): The ions gyro radius  $\rho$  for Nitrogen , Argon , and Hydrogen ions

ions	3*10 <sup>4</sup> Gaus	3.4*10 <sup>4</sup> Gaus	3.8*10 <sup>4</sup> Gaus	4.2*10 <sup>4</sup> Gaus	4.6*10 <sup>4</sup> Gaus	5*10 <sup>4</sup> Gaus	5.4*10 <sup>4</sup> Gaus
Ar	0.151265878	0.133469892	0.11942043	0.108047056	0.098651659	0.090759528	0.084036599
H <sub>2</sub>	43.28741766	38.19478028	34.1742771	30.91958404	2.823092456	25.97245059	24.04856536
N <sub>2</sub>	0.03421184	0.030186917	0.027009347	0.0244347028	0.022312069	0.020527104	0.019006577

(	Ibn Al-Haitham Journal for Pure and Applied Science						قية	لة و التطبي	م الصرف	الهيثم للعلو	مجلة إبن		
	No.	$\boxed{1}$	Vol.	25	Year	2012	π.	2012	السنة (	25	المجلد (	$\left(\begin{array}{c}1\end{array}\right)$	العدد

Table (9):The cross field diffusion coefficient  $D_{\rho}$  for Nitrogen , Argon , and Hydrogen ions

	(1/sec)							
Ar	38617.0166	30065.15117	24068.77766	19702.55959	16425.00687	13902.12634	11918.83231	
H <sub>2</sub>	3162424243	2462094998	1971040041	1613481757	1345076474	1138472727	976056864.7	Ś
N <sub>2</sub>	1975.375388	1537.921949	1231.189612	1007.844538	840.1900102	711.1351395	609.6837117	
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Ibn Al-Haitham Journal for Pure and Applied Science	مجلة إبن الهيثم للعلوم الصرفة و التطبيقية
No.         1         Vol.         25         Year         2012	العدد 1 المجلد 25 السنة 2012
لازما الممغنطة	انتقال الشحنة في البا

هادي جبار مجبل العجيلي ، دريد هاني يونس ، ايناس احمد جواد ، محسن عنيد حسوني قسم الفيزياء ، كلية التربية – ابن الهيثم ، جامعة بغداد

استلم البحث في :27 اذار 2011، قبل البحث في: 7 كانون الاول 2011

الخلاصة

الحلاصه مصدر البلازما يحدد حركة الشحنات في التوزيع غير المتوازن لطاقة الشحنة ويحد من انتقال الالكترونات عبر المجال الممغنط . حركة الالكترونات والشحنات مسيطر عليها بالمجال الكهربائي للتمدد الثنائي ايون – الكترون وتصادمات الشحنة . اعتمد مصدر الكثافة الواجب للالكترونات والايونات الحرارة اللازمة لحساب معامل الانتشار . معامل الانتشار الناشيء من تمدد الكترون ايون، ومعامل الانتشار للشحنة المنتقلة، ومعامل الانتشار التقريبي لتمدد ايونات الاكترون حسبت من خلال مجال البلازما المنتظم . اظهرت النتائج الحقيقية كيفية انحصار عمليات الانتشار من خلال خصائص البلازما .

الكلمات المفتاحية : - انتقال الشحنة ، البلازما الممغنطة ، معامل الانتشار